

Chemistry 333 Fall 2010
Organic Lab Lecture

Name: _____

Exam #2

This is an open-book, open notes exam. Show your work, so you can receive credit for correct parts of the final molecule.

1. (20 points) $C_6H_{12}O_3$

^{13}C NMR

205.2, s

101.5, d

53.8, q (2)

47.3, t

31.0, q

1H NMR

2.20, s, 3H

2.75, d, J=7.2 Hz, 2H

3.35, s, 6H

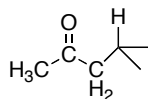
4.80, t, J=7.2 Hz, 1H

Step 2:



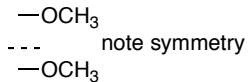
205.2, s = ketone

from generic chemical shifts:



2.20, s, 3H 2.75, d, J=7.2 Hz, 2H

Step 3: Two ethers. From generic chemical shifts,

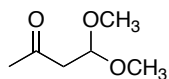


There is also



101.5, d

Putting it all together:



2. (40 points) $C_7H_{12}O$

^{13}C NMR

211.6, s

50.0, t

41.1, t

34.2, d

33.3, t

25.3, t

22.1, q

1H NMR

1.05, d, $J=7.3$ Hz, 3H

1.30, m, 1H

1.60, m, 1H

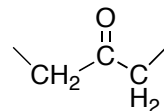
1.95, m, 3H

2.4, m, 4H

Step 2:



from generic chemical shifts:

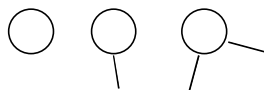


50.0, t
41.1, t

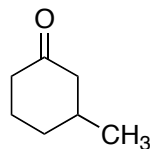
205.2, s = ketone

2.4, m, 4H

one ring



For every side chain attached to the ring, there must be one branching carbon and one end group. We have only one branching carbon, 34.2 d, and only one end group, 22.1, q, so the middle cartoon must represent the molecule. The methyl group is 1.05 d, so it must be directly attached to the ring



3. (40 points)

$C_{10}H_{16}O_2$

IR: 2932, 2854, 1825, 1132 cm^{-1}

^{13}C NMR

172.9, s

79.0, d

46.8, d

37.7, d

28.9, t (2)

28.1, t

25.0, t (2)

8.4, q

1H NMR

4.18, dd, $J = 10.6, 6.3$ Hz, 1H

3.72, qd, $J = 7.8, 6.3$ Hz, 1H

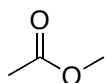
2.01-1.92, m, 1H

1.82-1.52, m, 5H

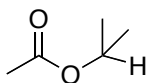
1.34, d, $J = 7.8$ Hz, 3H

1.32-1.19, m, 3H

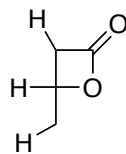
1.07-0.88, m, 2H



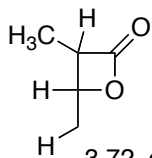
172.9, s is an ester



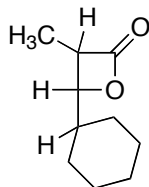
79.0, d tells us that there is one H where the O is attached



1825 cm^{-1} tells that it is a four-membered ring lactone. 4.18 dd tells us that there is one H on each side



3.72, dq, tells us that the H next to the carbonyl is also next to a CH_3



One more ring, with symmetry, and no additional branching